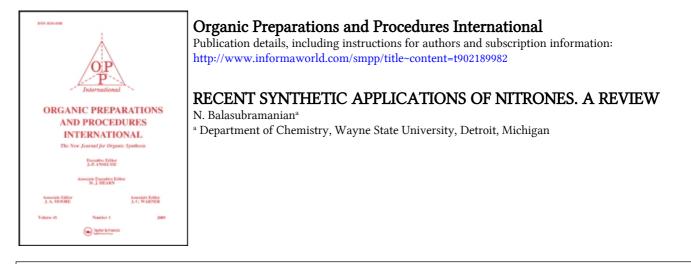
This article was downloaded by: On: *27 January 2011* Access details: *Access Details: Free Access* Publisher *Taylor & Francis* Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



To cite this Article Balasubramanian, N.(1985) 'RECENT SYNTHETIC APPLICATIONS OF NITRONES. A REVIEW', Organic Preparations and Procedures International, 17: 1, 23 – 47 To link to this Article: DOI: 10.1080/00304948509355467 URL: http://dx.doi.org/10.1080/00304948509355467

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

N. Balasubramanian

Department of Chemistry Wayne State University, Detroit, Michigan 48202

INTRODUCTION			25
I.	CYC	CLOADDITIONS	25
	1.	INTERMOLECULAR CYCLOADDITIONS	25
		a. β-Lactams	25
		b. Carbohydrates	27
		c. Amino Acids	28
		d. Alkaloids	29
	2.	INTRAMOLECULAR CYCLOADDITIONS	32
		a. C. Alkenyl Nitrones	33
		i. Sesquiterpenes	33
		ii. Alkaloids	34
		iii. Aminosugars	35
		iv. Miscellaneous	36
		b. N-Alkenyl Nitrones	37
		Alkaloids	38
11.	[4	+2] CYCLOADDITIONS	39
III	. от	THER SYNTHETIC APPLICATIONS	41
REF	EREN	ICES	43

[©]1985 by Organic Preparations and Procedures Inc.

N. Balasubramanian

Department of Chemistry Wayne State University, Detroit, Michigan 48202

INTRODUCTION

A highly useful reaction of nitrones is the [3+2] cycloadditon with olefins to afford five-membered heterocycles. This reaction, which results in the formation of carbon-carbon bond with the concomitant strategic 1,3-disposition of the nitrogen and oxygen functionalities, serves as an attractive alternative to the classic Mannich reaction. The general subject of nitrones¹ as well as the specific aspect of their preparations, rearrangments, participation in 1,3-dipolar cycloaddition reactions, their uses in medicinal chemistry, and in alkaloid synthesis have been reviewed up to the late seventies.² Mechanistic aspects of the cycloaddition reaction has been studied by different groups.³ The continuing interest in



the cycloaddition reactions of nitrones has prompted this review which emphasizes synthetic applications which have appeared since 1978.

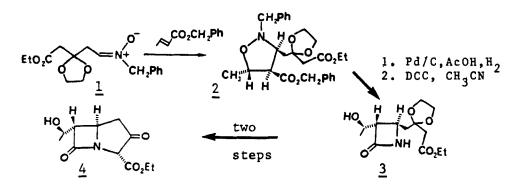
I. CYCLOADDITIONS

1. INTERMOLECULAR CYCLOADDITIONS

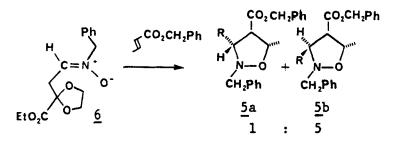
a. 8-Lactams

Thienamycin, a key member of the carbapenem antibiotics, was recently

synthesized by two groups; both employed a nitrone-olefin cycloaddition reaction.⁴ The predictable regio- and stereochemical outcome of this reac tion allowed control of stereochemistry at three centers around the azetidinone ring system. Kametani, *et al.* reported that isoxazolidine 2 was the sole isomer obtained from the cycloaddition of nitrone 1 and benzyl

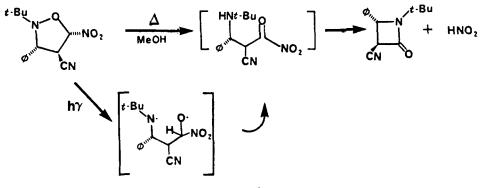


crotonate in benzene; an *endo*-crotonate-*E*-nitrone transition state was suggested. Isoxazolidine 2 was hydrogenated to a β -hydroxy- β -amino acid, which was cyclized to give 3. Similar results were reported for the cycloaddition of nitrone 6 and benzyl crotonate in toluene, except that two isomeric isoxazolidines were produced. The key intermediates 5a and 5b were transformed into the azetidinone system.



Thermal and photochemical reorganizations of isoxazolidines have resulted in the production of β -lactams.⁵ A ring contraction involving a reactive acyl nitro species is presumed to be involved (Scheme 1). In the

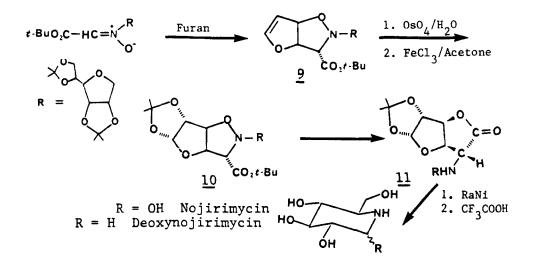
photolytic pathway, scission of the N-O bond followed by internal hydrogen atom transfer and cyclization was postulated. The presence of the nitro group was thought to be important.





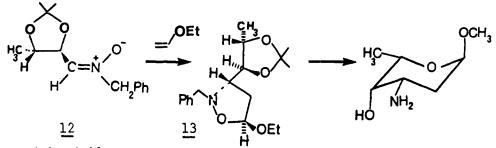
b. Carbohydrates

An interesting application of nitrone cycloaddition reaction is found in the total synthesis of nojirmycin, an antibiotic and one of the few naturally occurring aza analogs of glucose.⁶ Nojirimycin and 1-deoxynojirimycin were elaborated from the nitrone-furan adduct 9. Osmylation, protection followed by N-O bond cleavage and hydrolysis, gave the amino



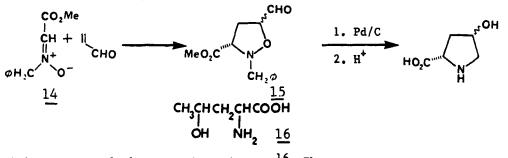
lactone 11, which was transformed in a few steps to the natural products. The nitrone employed in this case is derived from a sugar derivative; asymmetric induction with such nitrones has been studied by Vasella.⁷

A high degree of diastereoselectivity was observed in the cycloaddition of nitrone 12 with excess ethyl vinyl ether at 35°.⁸ The sole diastereomer 13 was subjected to N-O bond scission under catalytic hydrogenation in methanolic HCl to give the glycoside of daunosamine.



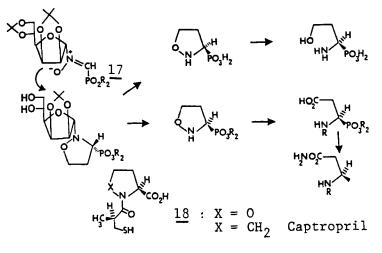
c. Amino Acids

The development of new methods for the preparation of amino acids continues to be a synthetic challenge. Cycloaddition of nitrone 14 with acrolein gave the isoxazolinidine 15. Hydrogenolysis over palladium hydroxide followed by hydrolysis produced the *cis*- and *trans*-4-hydroxyprolines.⁹ As expected a single regioisomer was formed as a stereoisomeric mixture. It was suggested that stereoisomers were the result of (*E*) and



(2) isomers of the starting nitrone.¹⁶ The same cycloaddition reaction served as the key step in the synthesis of γ -hydroxy- β -amino acid 16.

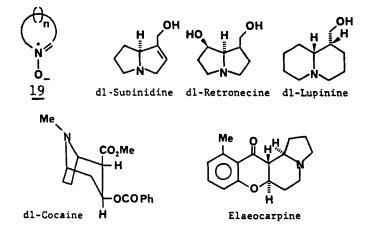
Asymmetric synthesis of α -aminophosphonic acids (analogs of α -amino acids) have been achieved by Vasella utilizing a new class of nitrones, C-phosphononitrones 17.¹⁰ The carbohydrate moiety attached to the nitrogen atom of the nitrone acts as a chiral auxiliary. Scheme 2 illustrates the synthesis of the α -aminophosphonic acid analogs of L-5-oxapro





line, L-homoserine, L-aspartic acid, L-asparagine. An analogous process was utilized in the synthesis of D- and L-5-oxaproline 18, a new captropril analogue.¹¹

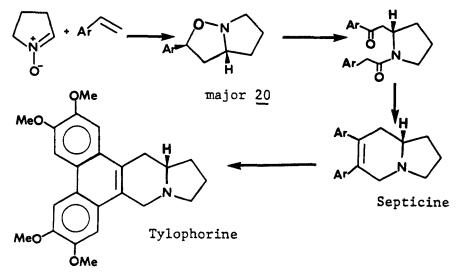
d. Alkaloids



Cyclic nitrones represented by the general formula **19** are commonly employed in the total synthesis of several classes of alkaloids. A recent account of the pioneerring work of Tufariello documents the diverse utility of these nitrones in the synthesis of senecio, indolizine, nuphar, and quinolizidine alkaloids.²

Indolizidine Alkaloids

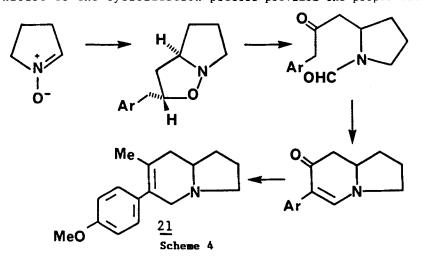
(\pm) Septicine and (\pm) tylophorine have a common parent system, aryl substituted indolizidine nucleus, and exhibit important biological activity. The key intermediate 20 required for the final photocyclization step was derived from a nitrone-olefin cycloaddition process (Scheme 3).¹²



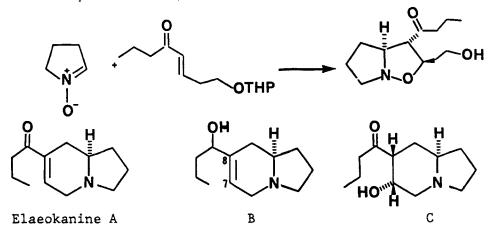
Scheme 3

The aglycone of ipalbine and ipomine alkaloids possess an indolizidine system with methyl substitution. Cycloaddition of p-allyl anisole and 1-pyrroline 1-oxide gave isoxazolidines,¹³ which was transformed to the N-formyl carbinol. Final cyclization to the bicyclic enamine was achieved with aluminum *t*-butoxide. The cyclized product was converted to the aglycone moiety 21 (Scheme 4).

Kametani reported¹⁴ the total synthesis of elaeokanine A,B,C, utilizing a nitrone-olefin cycloaddition. While the regio- and stereochemical nature of the cycloaddition process provided the proper basic

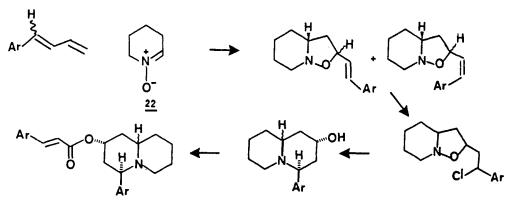


skeleton and correct stereochemistry at C-7 and C-8, the subsequent reaction sequence did not preserve the latter.



Quinolizidine Alkaloids

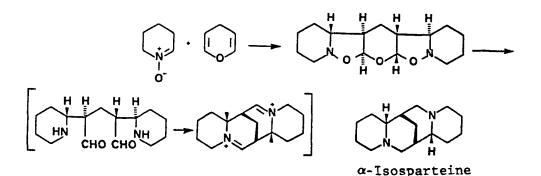
The regiospecific cycloaddition of (*E*) and (*Z*) dienes with nitrone 22 produced the isoxazolidines.¹⁵ Reduction of the N-O bond places the hydroxyl at the appropriate carbon. Further elaboration gave the alkaloids (\pm) -lasubine I, (\pm) -subcosine I. These systems were approached by Takano by similar manner.¹⁶



Subcosine

Lasubine

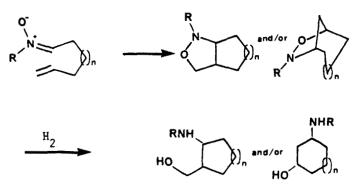
A short and facile preparation of the tetracyclic alkaloid, α -isosparteine, involves the cycloaddition of Δ^1 -piperidine-l-oxide to 4Hpyran.¹⁷ Catalytic hydrogenation of the adduct over Pd (OH)₂ gave the natural product. The stereochemical outcome in the cycloaddition resulted from the *exo*-addition product. No *endo*-product was observed due to the steric hindrance in the transition state of the *endo* mode of addition.



2. INTRAMOLECULAR CYCLOADDITIONS

Pioneering investigations of intramolecular nitrone-olefin cyclization were carried out by LeBel and coworkers in the late fifties and early sixties.¹⁸ In these reactions, the olefin moiety can be an appendage either from the carbon atom or from the nitrogen atom of the nitrone. In general, the isoxazolidines that are derived from C-alkenyl nitrones (with the exception of cyclic nitrones) are potential precursors for the

construction of carbocyclic systems (Scheme 5). On the other hand, the isoxazolidines derived from N-alkenyl nitrones will necessarily lead to nitrogen heterocycles (*vide infra*).

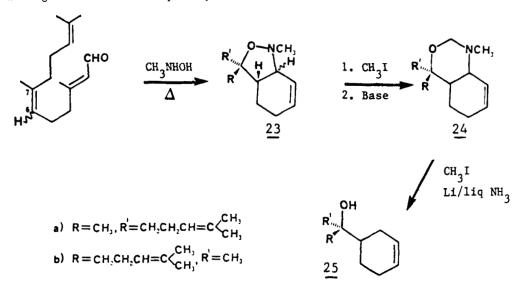




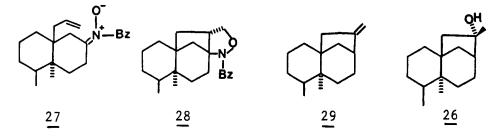
a. C-Alkenyl Nitrones

i) Sesquiterpenes

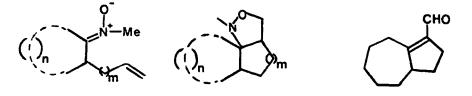
The stereospecificity of C-alkenyl nitrone cycloaddition reaction had been exploited for the synthesis of sesquiterpene bisabolol.¹⁹ The diastereomeric mixture of isoxazolidines 23a and 23b was obtained from (62) and (6E)-farnesal and methylhydroxylamine. Reductive removal of the nitrogen led to the sesquiterpene 25.



Intramolecular cycloadditions of exocyclic nitrones with the appropriate olefinic linkages furnished bridged bicycloalkanes fused to the isoxazolidines.²⁰ Application of this reaction is exemplified by the total synthesis of the sesquiterpene 26, (125)-7, 12-secoishwaran-12-ol. Nitrone 27 underwent smooth cycloaddition to give the key interme diate isoxazolidines 28 in high yields. Hydrogenation followed by hydrodeamination and dehydration gave the penultimate intermediate 29, which was converted to the natural product in two steps.



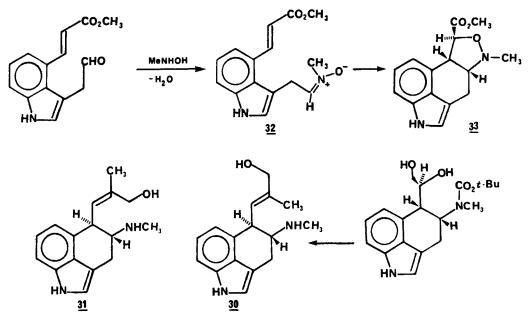
A related cycloaddition reaction was reported by Kakisawa and coworkers.²¹ Perhydroazulene derivative and related systems were derived from the resulting isoxazolidine (Scheme 6).



Scheme 6

ii) Alkaloids

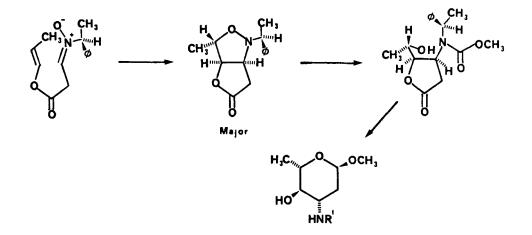
In a total synthesis of the ergot alkaloids chanaclavine I (30) and isochanaclavine (31) by Oppolzer, the key step involved a transient C-alkenylnitrone.²² Nitrone 32 underwent a regio- and stereoselective intra-molecular cycloaddition to a 1,2-disubstituted olefin. The isoxazoli-



dine **33** contains the proper basic skeleton for elaboration to the naturally occurring alkaloids.

iii) Aminosugars

Chiral synthesis of the sugar moities of the anthracycline antibiotics adriamycin, daunomycin, and carminomycin were achieved by Wovkulich and Uskokovic.²³ The nitrone-enol ester, generated by reaction of an enamine and optically pure α -methylbenzylhydroxylamine, underwent intra-

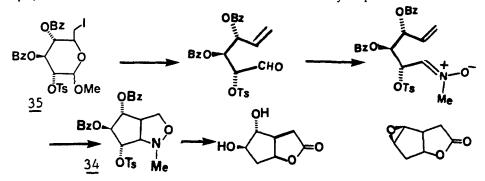


Scheme 7

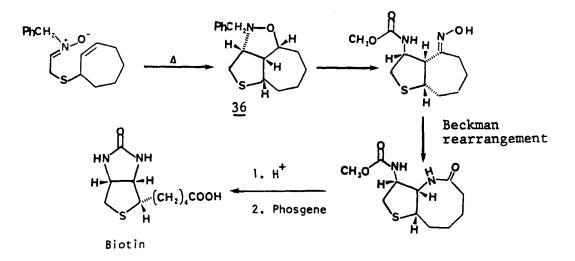
molecular cycloaddition to an adduct which has the desired relative stereochemistry at the chiral centers with appropriate heteroatoms in place. Reduction followed by protection of the amine gave a lactone carbamate, which was transformed to the sugar derivatives (Scheme 7).

iv) Miscellaneous

Isoxazolidine 34, obtained from the sugar derivative 35 in two steps,²⁴ was converted to various functionalized cyclopentane derivatives.



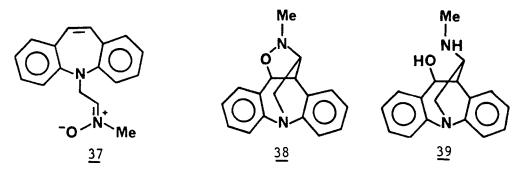
Noteworthy in these examples is the formation of the carbocycles with predictable stereochemistry at the various chiral centers. The starting carbohydrate nucleus served to introduce various functionalities and chirality in the carbocyclic system *via* the nitrone cycloaddition process.



Scheme 8

Isoxazolidines fused to heterocyclic system would be produced when a hetero atom links the cycloaddition partners.²⁵ Thus, in a single cyclo-addition process of **36**, there was obtained the necessary framework with proper stereochemistry at all three centers of biotin.²⁶ The side-chain and nitrogens for the urea moiety were derived from the seven-membered ring of the isoxazolidine (Scheme 8).

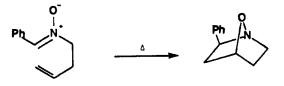
An interesting variation in these cycloaddition reactions involves the nitrone and olefin linked by a nitrogen atom.²⁷ Nitrones such as 37



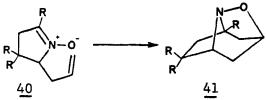
underwent cycloaddition to give the pentacyclic isoxazolidine 38 which in turn elaborated to functionalized dibenzazepines 39. The corresponding carbon analogs had also been synthesized.

b. N-Alkenyl Nitrones

Intramolecular reactions involving N-alkenyl nitrones which lead to bridged bicyclic isoxazolidines have attracted little attention in total synthesis until recently.²⁸ In the first example, reported by Lumma, N-3butenyl, C-phenyl nitrone cyclized to give the isoxazolidine exclusively.²⁹

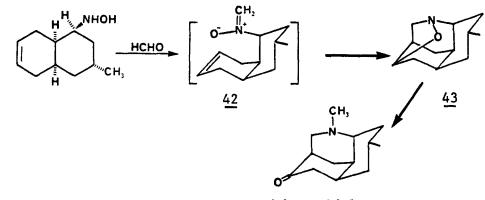


Analogous regiochemistry in the direction of addition was observed in the reaction of cyclic nitrone $40 \rightarrow 41$ leading to the tropane alkaloid skeleton.³⁰



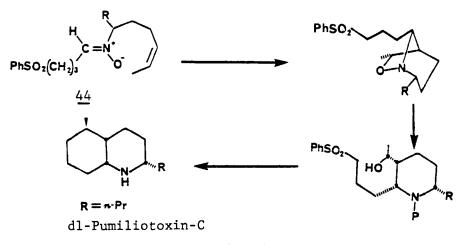
Alkaloids

An elegantly conceived total synthesis of Lycopodium alkaloid (+)luciduline from an N-alkenyl nitrone has been reported.³¹ A regiospecific intramolecular cycloaddition of nitrone 42, from (+)-pulegone, resulted in the exclusive formation of the isoxazolidine 43. Methylation followed by LAH reduction gave a N-methylamino alcohol which was oxidized to (+)luciduline.



(+)-Luciduline

N-Alkenyl nitrones of the type **44** undergo smooth cyclization to the bridged bicyclic isoxazolidines; N-O bond scission lead to 2,3,6-*cis* substituted piperidines.³² The relative stereochemistry of four chiral centers are controlled in a single cycloaddition process. This methodology was extended to the total synthesis of *dl*-pumiliotoxin-C from 2-pentanone and methyl phenyl sulfone (Scheme 9).

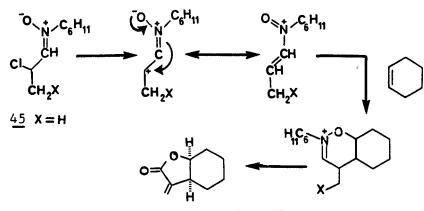


Scheme 9

II. [4+2] CYCLOADDITIONS OF NITRONES

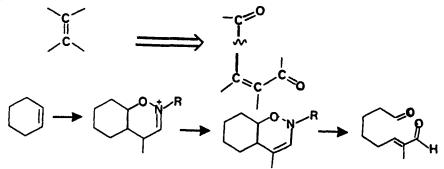
C-(1-chloroalkyl) nitrones, (e.g., 45) introduced by Eschenmoser,³³ can be efficiently utilized in organic synthesis. These nitrones serve as the precursors to the N-alkyl-N-vinyl nitrosonium cations, which participate in [4+2] cycloaddition process. This methodology was artfully applied in the synthesis of α -methylene lactones (Scheme 10) and in other useful synthetic transformations.

The sequence [2+4]-cycloaddition→deprotonation→[2'+4']cycloreversion was employed for the oxidative cleavage of olefinic double bonds with concomitant extension of the carbon chain at one of the double bond termini.

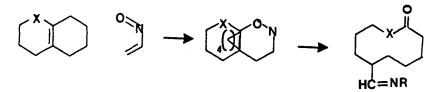


Scheme 10

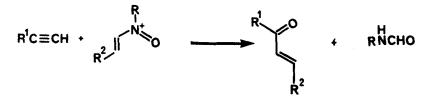
This indirect "carboxolytic" cleavage of double bonds proceeds in preparatively useful yields.³³



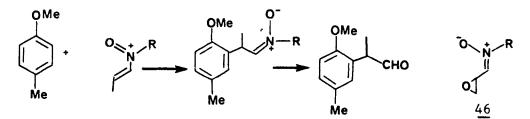
An extension of this methodology involving enol ethers as the cycloaddition partners provided a synthetic route to medium ring lactones. However, it is observed that these enol ether type double bonds cause difficulties in the initial cycloaddition process. Also the delicate cycloreversion process is limited by steric effects.³⁴



This cycloaddition, reversion process is also applied to acetylinic compounds to provied α , β -unsaturated carbonyl compounds.³⁵



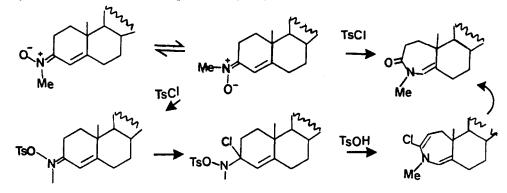
Simple variation in the reaction conditions of argentous ion induced α -chloro-aldonitrone/olefin reaction proceeds in substitution, thereby providing a method for β -y-unsaturated aldehyde. Such substitution also occurs at activated aromatic nuclei. These reactions occur regiospecifically at the least substituted carbon.³⁶ Similar reactivity was observed for the epoxy nitrones (46).³⁷



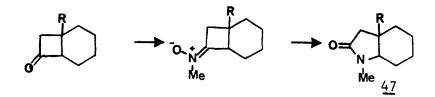
III. OTHER SYNTHETIC APPLICATIONS

Barton-Beckmann Rearrangement

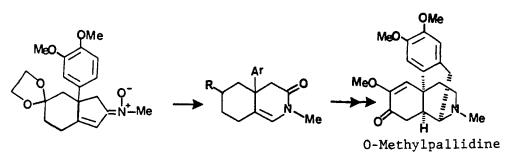
Ketonitrones undergo a Beckmann-type rearrangement on treatment with p-toluenesulfonyl chloride.³⁸ The outcome of this reaction, in contrast to the Beckmann rearrangement, does not depend on the stereochemistry of the nitrone since the (2)-and (E)-nitrones interconvert rapidly under the conditions and both gave the same product. Such a rearrangement of bicyclo [4.2.0] octanones gave the perhydroindolone 47, while the conven-



tional Beckmann rearrangement would only lead to isoindole skeleton. 39



More recently, McMurry applied this rearrangement in his synthesis of O-methylpallidinine. 40

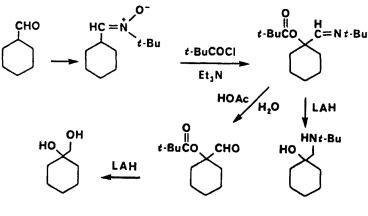


a-Oxygenation of Aldehydes and Cyclic Ketones

This process involves the treatment of the nitrones with acyl chloride in the presence of triethylamine to afford α -acyloxyimines by rearrangement of N-vinyl-O-acylhydroxylamine intermediates.⁴¹ Hydrolysis yields the corresponding α -acyloxy carbonyl compounds (Scheme 11).

Other useful synthetic transformations $^{42-46}$ which need to be mentioned include the formation of olefins, 42 oxaziridines, 43 oxime-0-ethers 44 and oxazines. 45

It is apparent that the nitrone-olefin cycloaddition process can serve as a powerful synthetic tool, and its full potential for diverse utility in synthesis is yet to be realized.



Scheme 11

Acknowledgment.- The author wishes to express his sincere thanks to Professors Carl R. Johnson and Norman A. LeBel for their help during the preparation of this manuscript.

REFERENCES

- L. I. Smith, Chem. Rev., <u>23</u>, 193 (1938); J. Hamer and A. Macaluso, ibid., <u>64</u>, 473 (1964); G. R. Delpierre and M. Lamchen, Quart. Rev., <u>19</u>, 329 (1965); P. A. S. Smith "Open-Chain Nitrogen Compounds," Vol. 2, p. 29, W. A. Benjamin, New York, 1966; M. Lamchen in "Mechanisms of Molecular Migrations," Vol. 1, p. 1, p. 1, B. S. Thyagarajan, Wiley-Interscience, New York, 1968; W. Rundel in Houben-Weyl: "Methoden der Organischen Chemie," Vol. 10/4, p. 311, E. Muller, Georg Thieme Verlag, Stuttgart, 1968; H. Stamm in "Methodicum Chimicum," Vol. 6, p. 333, F. Zymalkowski, Georg Thieme Verlag, Stuttgart, 1975; S. R. Sandler and W. Karo, "Organic Functional Group Preparations," Vol. III, p. 301, Academic Press, New York, 1972; Eli Breuer in "The Chemistry of Functional Groups," (Supplement F), Part 1, p. 459, S. Patai, John Wiley & Sons, New York, 1982.
- D. St. C. Black, R. T. Crozier and V. C. Davids, Synthesis, 205 (1975); J. J. Tuffariello, Accounts Chem. Res., <u>12</u>, 396 (1979); W. Kliegel, Pharmazie, <u>32</u>, 643 (1977), <u>33</u>, 331 (1978).
- G. Bianchi, C. DeMicheli and R. Gaudolfi in Supplement A: "The Chemistry of Double-Bonded Functional Groups," p. 369, S. Patai, John Wiley and Sons, London, 1977; R. Huisgen J. Org. Chem., <u>41</u>, 403 (1976); R. A. Firestone, Tetrahedron, <u>33</u>, 3009 (1977); R. D. Harcourt, ibid., <u>34</u> 3125 (1978); Y. M. Chang, J. Sims and K. N. Houk, Tetrahedron Lett., 4445 (1975); D. St. C. Black, R. F. Crozier and I. D. Rae, Australian J. Chem., <u>31</u>, 2239 (1978); R. Huisgen, Angew. Chem. Int. Ed., <u>2</u>, 565, 633, (1963); R. Huisgen, J. Org. Chem., <u>33</u>, 2291 (1968); R. Huisgen, R. Sustmann and K. Bunge, Chem. Ber., <u>105</u>, 1324 (1972); K. N. Houk, Accounts Chem. Res., <u>8</u>, 361 (1975); D.

Mukherjee, L. S. Dommelsmith and K. N. Houk, J. Am. Chem. Soc., <u>100</u>, 1954 (1978); M. Joucla and J. Hamelin, J. Chem. Res., 276 (S), 3535 (M) (1978).

- T. Kametani, S. P. Huang, A. Nakayama and T. Honda, J. Org. Chem., <u>47</u>, 2328 (1982); R. V. Stevens and K. Albizali, Chem. Commun., 104 (1982).
- 5. A. Padwa, K. F. Koehler and A. Rodiquez, J. Am. Chem. Soc. <u>103</u>, 4974 (1981); for the photolytic conversion of N-methyl-isoxazolidine to oxazine, see N. A. LeBel, T. A. Lajiness and D. B. Ledlie, J. Am. Chem. Soc., 89, 3076 (1967).
- 6. A. Vasella and R. Voeffary, Helv. Chim. Acta., <u>65</u>, 1134 (1982).
- 7. A. Vasella, ibid., <u>60</u>, 426, 1273 (1977).
- 8. P. DeShong and J. M. Loginus, J. Am. Chem. Soc. 105, 1606 (1983).
- 9. J. Hara, Y. Inouye and H. Kakisawa, Bull. Chem. Soc., Japan, <u>54</u> 3871 (1981); Y. Inouye, Y. Watanabe, S. Takahashi, H. Kakisawa, ibid., <u>52</u>, 3763 (1979).
- 10. A. Vasella and R. Voeffray, Helv. Chim. Acta., 65, 1953 (1982).
- A. Vasella, R. Voeffray, J. Pless and R. Huguenin, ibid., <u>66</u>, 1241 (1983).
- 12. H. Iida, M. Tanaka and C. Kibayashi, Chem. Commun., 271 (1983).
- 13. H. Iida, Y. Watanabe and C. Kibayashi, Chem. Lett., 1195 (1983).
- H. Otomasu, N. Takatsu, T. Honda and T. Kametani, Tetrahedron, <u>38</u>, 2627 (1982).
- 15. H. Iida, M. Tanaka and C. Kibayashi, Chem. Commun., 1143 (1983).
- 16. S. Takano and K. Shishido, ibid, 940 (1983).
- 17. H. Oinuma, S. Dau and H. Kakisawa, ibid., 654 (1983).

- 18. A. C. Cope and N. A. LeBel, Abs. of Papers, 133rd Meeting, American Chemical Society, p. 62-N. San Francisco, April 13-18 (1958); A. C. Cope and N. A. LeBel, J. Am. Chem. Soc., <u>82</u>, 4656 (1960); N. A. LeBel, Trans. N. Y. Acad. Sci., <u>27</u>, 858 (1965).
- 19. M. A. Schwartz and G. C. Swanson, J. Org. Chem., 44, 953 (1979).
- R. L. Funk, L. H. M. Horcher, II, J. U. Daggett and M. M. Hansen, ibid., <u>48</u>, 2632 (1983).
- S. Takahashi, T. Kusumi, Y. Sato, Y. Inouye and H. Kakisawe, Bull. Chem. Soc., Japan, 54, 1777 (1981).
- W. Oppolzer and J. I. Grayson, Helv. Chim. Acta., <u>63</u>, 1706 (1980); W. Oppolzer, J. I. Grayson, H. Wegmann and M. Urrea, Tetrahedron, <u>39</u>, 3695 (1983).
- P. M. Wovkulich and M. R. Uskokovic, J. Am. Chem. Soc., <u>103</u>, 3956 (1981).
- 24. R. J. Ferrier and R. H. Furneaux, et al., J. Chem. Soc., Perkin Trans I., 1623 (1983).
- W. Oppolzer and K. Keller, Tetrahedron Lett., 4313 (1970); W.
 Oppolzer and H. P. Weber, ibid., 1121 (1970); W. Oppolzer and K.
 Keller, ibid., 1117 (1970); R. Brambilla, R. Friary, A. Ganguly, M.
 S. Puar, B. R. Sunday and J. J. Wright, Tetrahedron, <u>37</u>, 3615 (1981).
- 26. P. N. Confalone, P. Pizzolato, D. L. Confalone and M. R. Uskokovic,
 J. Am. Chem. Soc., <u>102</u>, 1954 (1980); P. N. Confalone, E. D. Lollar,
 G. Pizzolato and M. R. Uskokovic, ibid., <u>100</u>, 6291 (1978).
- 27. P. N. Confalone and E. Huie, J. Org. Chem., 48, 2994 (1983).
- 28. W. Oppolzer, Agnew. Chem. Int. Ed., 16, 10 (1977).
- 29. W. C. Lumma, Jr., J. Am. Chem. Soc., 91, 2820 (1969).
- 30. J. J. Tufariello and G. B. Mullen, ibid., 100, 3638 (1978).

2011

Downloaded At: 11:27 27 January

- 31. W. Oppolzer and M. Petrzilka, Helv. Chim. Acta., <u>61</u>, 2755 (1978). 32. N. A. LeBel and N. Balasubramanian, Abs. of Papers, 186th Meeting, American Chemical Society, No. 123 (1983); N. Balasubramanian, Ph.D. Dissertation, Wayne State University, 1983.
- 33. U. M. Kempe, T. K. Das Gupta, K. Blatt, P. Gygax, D. Felix and A. Eschenmoser, Helv. Chim. Acta., <u>55</u>, 2187 (1972); T. K. Das Gupta, D. Felix, U. M. Kempe and A. Eschenmoser, ibid., <u>55</u>, 2198 (1972); M. Petrzilka, D. Felix and A. Eschenmoser, ibid., <u>56</u>, 2950 (1973); P. Gygax, T. K. Das Gupta and A. Eschenmoser, ibid., <u>55</u>, 2205 (1972).
- E. Shalom, J. L. Zevon and S. Shatzmiller, J. Org. Chem., <u>42</u>, 4213 (1977).
- S. Shatzmiller and A. Eschenmoser, Helv. Chim. Acta., <u>56</u>, 2975 (1973).
- S. Shatzmiller, P. Gygax, D. Hall and A. Eschsenmoser, ibid., <u>56</u>, 2961 (1973).
- 37. M. Riediker and W. Graf, ibid., 62, 205 (1979).
- D. H. R. Barton, M. J. Day, R. H. Hesse and M. M. Pechet, J. Chem. Soc., Perkin I, 1764 (1975).
- 39. P. W. Jeffs and G. Molina, Chem. Commun., 3, (1973).
- 40. J. E. McMurry and V. Farina, Tetrahedron Lett., <u>24</u>, 4653 (1983).
- 41. C. H. Cummins and R. M. Coates, J. Org. Chem., <u>48</u>, 2070 (1983).
- 42. D. R. Boyd, D. C. Neill and M. E. Stubbs, J. Chem. Soc., Perkin II, 30 (1978); D. R. Boyd and D. C. Neill, J. Chem. Soc., Perkin I, 1309 (1977); W. M. Leyshon and D. A. Wilson, ibid., 1920 (1975); D. R. Boyd, Tetrahedron Lett., 1467 (1973); H. K. Kim and P. M. Weintraub, J. Org. Chem., <u>35</u>, 4282 (1970).

- 43. J. P. Alazard, B. Khemis and X. Lusinchi, Tetrahedron, <u>31</u>, 1427 (1975); D. St. C. Black, N. A. Blackman and A. B. Boscacci, Tetrahedron Lett., 175 (1978); Australian J. Chem., <u>32</u>, 1775 (1979);
 W. M. Leyshson and D. A. Wilson, J. Chem. Soc., Perkin I, 1925 (1975).
- 44. T. Kusmui, K. Yoneda and H. Kakisawa, Synthesis, 221 (1979).
- 45. D. Hwang, M. E. Post and N. A. LeBel, J. Org. Chem., 44, 1819 (1979).
- 46. J. E. Baldwin, M. F. Chan, G. Gallacher, P. Monk and K. Prout, Chem. Commun., 250 (1983); T. Koizumi, H. Hirai and E. Yoshii, J. Org. Chem., <u>47</u>, 4004 (1982); C. Belzecki and I. Panfil, ibid., <u>44</u>, 1212 (1979); P. M. Wovkulick, F. Barcelos, A. D. Batcho, E. G. Baggiolini and M. R. Uskokovich in Abs. of Papers, 186th Meeting, American Chemical Society, No. 201 (1983); W. R. Roush and A. E. Walts, J. Am. Chem. Soc., <u>106</u>, 721 (1984); C. M. Tice and B. Ganem, J. Org. Chem., <u>48</u>, 5048 (1983), P. Storme, P. Callant and M. Van de Walle, Bull. Soc. Chim., Belges., <u>92</u>, 1019 (1983).

(Received January 6, 1984; in revised form October 19, 1984)